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Electron tunneling in ruthenium-modified cytochrome *c*

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Distant Fe^{2+} - Ru^{3+} electronic couplings have been extracted from intramolecular electron-transfer rates in $\text{Ru}(2,2'$ -bipyridine)₂(imidazole)(histidine-*X*)²⁺ (*X* = 33,39,62,72,79) derivatives of cytochrome *c*. The rates are $> 1 \cdot 10^8$ (79); $3.2(4) \cdot 10^6$ (39); $2.6(3) \cdot 10^6$ (33); $9.0(3) \cdot 10^5$ (72); $1.0(2) \cdot 10^4$ s⁻¹ (62); the couplings increase according to 62 (0.006) < 72 (0.057) < 33 (0.097) < 39 (0.11) < 79 (> 0.6 cm⁻¹). The rates (and the couplings) correlate with the lengths of σ -tunneling pathways comprised of covalent bonds, hydrogen bonds, and through-space jumps from the histidines to the heme group.

The electron-transfer (ET) reactions that occur within and between proteins typically involve prosthetic groups separated by large molecular distances (> 10 Å). An understanding of how the intervening medium, driving force, and nuclear reorganization energetics and dynamics modulate these long-range protein ET reactions has been a central goal in this field [1–22]. Of particular interest has been the mechanism by which the peptide matrix promotes electronic coupling between distant redox sites [4–10]. Our investigations of electron transfer in ruthenium-modified proteins have suggested that the structure of the peptide between the donor and acceptor controls the coupling [1,12,16–20].

We have studied intramolecular electron transfer by attaching photoactive Ru complexes to protein surfaces [17]. $\text{Ru}(\text{bpy})_2(\text{CO}_3)$ has been shown to react with surface His residues to yield, after addition of excess imidazole (im), $\text{Ru}(\text{bpy})_2(\text{im})(\text{His})^{2+}$ [21]. The protein-bound Ru complexes are luminescent, but the excited states ($^*\text{Ru}^{2+}$) are rather short lived ($\tau \leq 100$ ns). When direct ET from $^*\text{Ru}^{2+}$ to the heme ($^*k_{\text{ET}}$) cannot compete with excited-state decay (k_{D}), ET quenchers (e.g., $\text{Ru}(\text{NH}_3)_6^{3+}$) are added to the solution to intercept a small fraction (1–10%) of the excited molecules, yielding (with oxidative quenchers) Ru^{3+} . If, before laser excitation of the Ru site, the heme is reduced, then $\text{Fe}^{2+} \rightarrow \text{Ru}^{3+}$ ET ($^b k_{\text{ET}}$) can be monitored by transient absorption spectroscopy. We have

used this method to examine ET rates in five different modified cytochromes: $\text{Ru}(\text{His}33)$, $^b k_{\text{ET}} = 2.6(3) \cdot 10^6$ s⁻¹; $\text{Ru}(\text{His}39)$, $3.2(4) \cdot 10^6$ s⁻¹; $\text{Ru}(\text{His}62)$, $1.0(2) \cdot 10^4$ s⁻¹; $\text{Ru}(\text{His}72)$, $9.0(3) \cdot 10^5$ s⁻¹; $\text{Ru}(\text{His}79)$, $> 10^8$ s⁻¹. Only in $\text{Ru}(\text{His}79)$ did $^*\text{Ru}^{2+} \rightarrow \text{Fe}^{3+}$ ET measurably accelerate excited-state decay; the rates for this reaction in the four other $\text{Ru}(\text{bpy})_2$ -modified proteins were determined from the small yields of Ru^{3+} - Fe^{2+} detected by transient absorption spectroscopy: $\text{Ru}(\text{His}33)$, $^*k_{\text{ET}} = 2(1) \cdot 10^5$ s⁻¹; $\text{Ru}(\text{His}39)$, $1.4(5) \cdot 10^6$ s⁻¹; $\text{Ru}(\text{His}62)$, $1.1(2) \cdot 10^5$ s⁻¹; $\text{Ru}(\text{His}72)$, $3.4(7) \cdot 10^5$ s⁻¹; $\text{Ru}(\text{His}79)$, $> 5 \cdot 10^7$ s⁻¹ [20].

According to semiclassical ET theory, rates become activationless when the reaction driving force ($-\Delta G^\circ$) equals the reorganization energy (λ) [2]. The driving force (0.74 eV) is approximately equal to the reorganization energy (0.8 eV) estimated for the $\text{Ru}(\text{bpy})_2(\text{im})(\text{His})$ -cyt *c* reactions [18]. The activationless (maximum) rates are limited by an electronic factor, $k_{\text{max}} = (\pi/\hbar^2 \lambda k_{\text{B}} T)^{1/2} H_{\text{AB}}^2$, where H_{AB} is the matrix element that couples the reactants and products at the transition state. Values of k_{max} and H_{AB} for the $\text{Fe}^{2+} \rightarrow \text{Ru}^{3+}$ ET reactions are given in Table I.

TABLE I

Electron-transfer parameters for $\text{Ru}(\text{bpy})_2(\text{im})(\text{HisX})$ -cytochromes *c*

<i>X</i>	[Fe^{2+} - Ru^{3+}]			
	k_{max} (s ⁻¹)	H_{AB} (cm ⁻¹)	<i>d</i> (Å)	$\sigma \angle$ (Å)
79	$> 1.0 \cdot 10^8$	> 0.6	4.5	11.2
39	$3.3 \cdot 10^6$	0.11	12.3	19.6
33	$2.7 \cdot 10^6$	0.097	11.1	19.5
72	$9.4 \cdot 10^5$	0.057	8.4	24.6
62	$1.0 \cdot 10^4$	0.006	14.8	28.8

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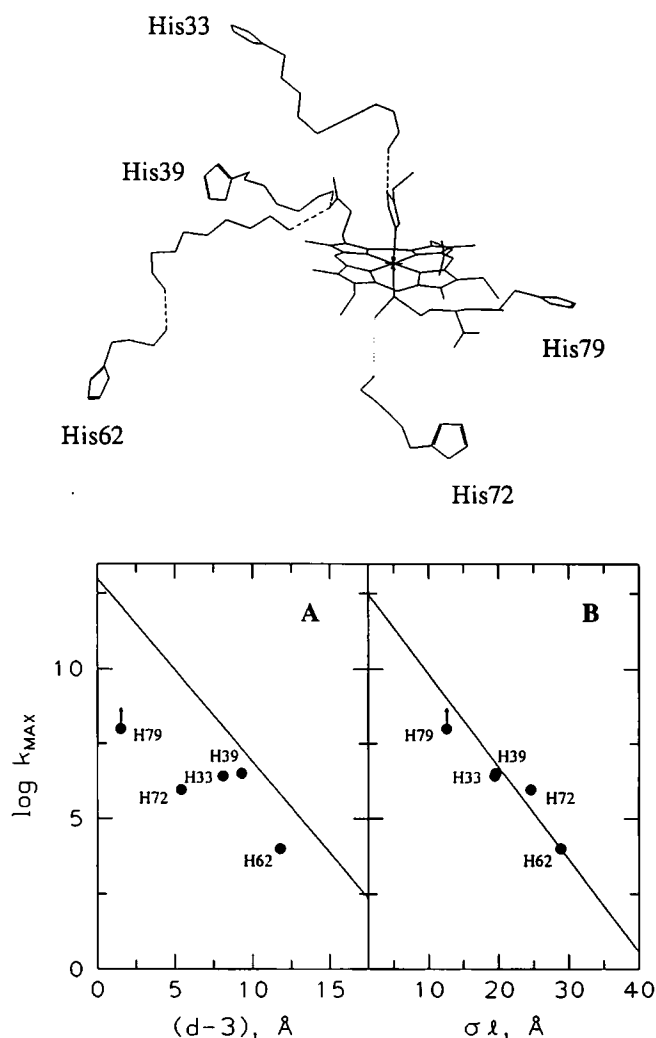


Fig. 1. Top: electronic-coupling pathways to the heme from Ru-modified residues in cytochrome *c*. (A) Maximum ET rates vs. d minus 3 Å (van der Waals contact). Exponential-decay line with $1 \cdot 10^{13} \text{ s}^{-1}$ intercept and 1.4 Å^{-1} slope. (B) Maximum ET rates vs. σl : 0.71 Å^{-1} slope; $3 \cdot 10^{12} \text{ s}^{-1}$ intercept.

Activationless intraprotein ET spanning 12 orders of magnitude in rate and nearly 19 Å in redox-site separation has been interpreted in terms of edge-edge distance (d) exponential decay with 1.4 Å^{-1} slope and $1 \cdot 10^{13} \text{ s}^{-1}$ intercept [22]. Implicit in this interpretation is the assumption that the intervening polypeptide can be treated as a homogeneous medium. This correlation, then, serves as a reference line; deviations from this line indicate situations in which inhomogeneities of the intervening peptide must be considered. Since at least one ($X = 72$) of our k_{\max} values falls over three orders of magnitude below this edge-edge exponential-decay line (there will be two rates in this category if, as is likely, the $X = 79$ k_{\max} is below 10^{10} s^{-1}) (Fig. 1A), and two others ($X = 33, 62$) deviate from the line by more than a factor of 50, analysis in terms of the structure of the intervening medium is called for.

Several models that take into account the inhomogeneity of the protein have been developed [5–11]. Beratan and Onuchic describe the coupling between redox centers in a protein in terms of pathways comprised of covalent, H-bonded, and through-space contacts [5,6]. An algorithm has been developed that searches a protein structure for the best pathways coupling two sites in a protein (the pathways between the histidines (33,39,62,72,79) and the heme are shown in Fig. 1). A given coupling pathway consisting of covalent, H-bonded, and through-space links can be described in terms of an equivalent covalent pathway with an effective number of covalent bonds (n_{eff}). Multiplying the effective number of bonds by 1.4 Å/bond gives σ -tunneling lengths (σl) for the five pathways (Table I) that correlate well with the maximum ET rates (one-bond limit set at $3 \cdot 10^{12} \text{ s}^{-1}$; slope of 0.71 Å^{-1}) (Fig. 1B). The 0.71 Å^{-1} decay accords closely with related distance dependences for covalently coupled donor-acceptor molecules [23].

We conclude that the structure of the intervening medium influences distant electronic couplings in cytochrome *c*. Importantly, Gruschus and Kuki have made inhomogeneous-aperiodic-lattice-model calculations on three derivatives ($X = 33, 39, 62$) that are in good agreement with our experimentally derived values (Gruschus, J.M. and Kuki, A., unpublished data).

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